## STEREOSPECIFIC DISPLACEMENTS OF p-toluenesulfonate and iodide ions with organolithium reagents

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Several recent studies have shown that Wurtz type coupling reactions between alkyllithium reagents and alkylhalides proceed by two one-electron transfer processes rather than a direct nucleophilic displacement.<sup>1</sup> However, certain charge delocalized organolithium reagents displace the bromide and chloride leaving groups from secondary carbon with a high degree of stereospecificity indicating typical  $S_n^2$  type displacement.<sup>2</sup>

We have found that allyllithium also displaces both the p-toluenesulfonate and iodide leaving groups from secondary carbon with a high degree of stereospecificity, 93% and 90% inversion of configuration, respectively.<sup>3</sup> 2-Octyl tosylate was prepared from (-)2-octanol,  $[\alpha]_D$  -8.0° (neat), and treated with allyllithium in ethyl ether solvent at 20° C. to give (-)4-methyl-1-decene,  $[\alpha]_D$  -5.25° (neat),  $n_D^{25}$  1.4246 (Lit.<sup>6</sup>  $[\alpha]_D$ -7.0° (neat),  $n_D^{25}$  1.4246) in at least 60% yield.<sup>7</sup> (+)2-Iodooctane,  $\alpha_D$  +45.8° (neat), coupled with allyllithium to give (+)4-methyl-1-decene,  $[\alpha]_D$ +4.5° (neat),  $n_D^{25}$  1.4244, in at least a 65% yield.<sup>7</sup>

 $(-)^{2}-C_{8}H_{17}OH \xrightarrow{CH_{3}C_{6}H_{4}SO_{2}C1} (-)^{2}-CH_{3}C_{6}H_{4}SO_{3}C_{8}H_{17} \xrightarrow{CH_{2}=CHCH_{2}Li} (-)^{2}-CH_{3}C_{6}H_{4}SO_{3}C_{8}H_{17} \xrightarrow{CH_{17}OH} (-)^{2}CH_{13}CH(CH_{3})CH_{2}CH=CH_{2}$ 

The relative configuration of 2-octyl tosylate, 2-iodooctane, and 4-methyl-1-decene can be rigorously shown. (-)2-Bromooctane has been related to (-)4-methyl-1-decene,<sup>6</sup> and

(+)2-bromooctane, (+)2-iodooctane, and (+)2-octanol have the same configuration.<sup>8</sup> Therefore, both reactions with allyllithium must proceed with inversion of configuration.

The high degree of stereospecificity of these reactions is expected if the coupling reactions are proceeding by an  $S_N^2$  pathway involving a rate controlling transition state similar to that suggested by Eastham and Gibson.<sup>9</sup>

A similar high degree of stereospecificity with yields of at least 60% was also observed for the reactions of another charge-delocalized organolithium reagent, benzyllithium, with 2-octyl tosylate and 2-iodooctane. All charge delocalized organolithium reagents, however, do not appear to react with such high yields with tosylates since Meyers and coworkers<sup>10</sup> indicate that the yield of coupled product for the reaction of the lithio salt of 2,4,4,6-tetramethyl-5,6-dihydro-1,3(4H)-oxazine with butyl tosylate was very low. The good yields of coupled products make allyllithium an attractive reagent for extending carbon chains by three carbons as well as adding a versatile functional group.

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## References

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- 3. The stereospecificity of these reactions was based on the following maximum rotations of reactants and products: 2-octanol,  $[\alpha]_D 9.9^\circ$  (neat),<sup>4</sup> 2-iodooctane,  $\alpha_D 64.2^\circ$  (neat),<sup>5</sup> and 4-methyl-1-decene,  $[\alpha]_D 7.0^\circ$  (neat).<sup>6</sup>
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